

[54] PRODUCTION OF ALUMINA FROM ORES

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[58] Field of Search 423/126, 132, 489, 483

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[57] ABSTRACT

An aluminum-containing ore such as a kaolin or montmorillonite clay is treated with a fluorine acid such as HF or H₂SiF₆ to produce aluminum fluoride (AlF₃) which is recovered from the liquid phase of the reaction mixture by crystallization as AlF₃ · 3H₂O crystals which are then dried and dehydrated to yield AlF₃. The crystallization mother liquor is recycled to the process to recover its AlF₃ values. The AlF₃ · 3H₂O crystals and separated solid phase of the reaction mixture are washed to recover AlF₃ values and the wash streams are recycled to the process. To produce alumina, the AlF₃ is pyrohydrolyzed to Al₂O₃ and HF. The HF is absorbed in the crystallization mother liquor and recycled with the mother liquor to the ore treatment step. All or a part of fluorine acid requirements can also be supplied from any suitable source such as, for example, a wet process phosphoric acid facility which is a plentiful source of inexpensive fluorine. The fluorine acid requirements can be supplied in a recycle stream of the process such as the crystallization mother liquor and/or a fresh make-up stream. The process converts the aluminum values in aluminum-containing ore deposits to a form (Al₂O₃) useful in the production of aluminum metal.

8 Claims, 3 Drawing Figures

